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hint s, k_m

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 $2 \text{ICl} + \text{H}_2 \rightarrow 2 \text{HI} + 2 \text{HCl}$
rate $k [\text{ICl}]^2 [\text{H}_2]$
a chemist proposes a mechanism for the reaction that is consistent with the rate law the mechanism has two elementary steps and the first step is slow compared to the second

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